Tetrahedron Letters, Vol.26, No.25, pp 2997-3000, 1985 0040-4039/85 \$3.00 + .00 Printed in Great Britain @1985 Pergamon Press Ltd.

STEREOCONTROL IN ISOXAZOLINE REDUCTIONS. SYNTHESIS OF AMINODEOXY-DL-XYLO- AND ARABINO-PENTOSE DERIVATIVES FROM FUROISOXAZOLINES ¹

Volker Jäger ^{*} and Ingrid Müller Institut für Organische Chemie der Universität Würzburg, Am Hubland D-8700 Würzburg

> Erich F. Paulus Hoechst AG, D-6230 Frankfurt a.M. 80

Summary: Ozonolysis of furoisoxazolines furnishes cis-4-oxygenated isoxazolines, which by suitable derivatization and subsequent LAH reduction yield xylo or arabino aminodeoxy-pentoses, depending on the presence of a free or blocked 4-hydroxy group.

Isoxazolines are useful intermediates in short, stereoselective syntheses of diverse natural and unnatural products,² notably amino compounds (by reduction).¹⁻⁵ The stereoselectivity of isoxazoline reductions effected by LAH is primarily determined by the nature, position, and number of substituents of C-4 and C-5 of the heterocyclic nucleus.⁵⁻⁷ While most groups that have been checked so far are <u>anti</u>-directing in this respect, free 4-hydroxyl induces <u>syn</u> hydride addition.^{3,7} Together with the remarkable additivity of substituent directing effects ($\Delta\Delta G^{\dagger}$ values),⁶ this permits <u>stereoselective</u> access to a large variety of γ -amino alcohols (including amino polyols, amino sugars and amino acids) in a predictable way.⁵ The present letter deals with the following questions on the feasibility of exerting <u>stereocontrol</u> in these reductions: (i) does the strong <u>syn</u>-directing effect of 4-OH prevail also in the presence of other <u>cis</u>-substituents? (ii) is it possible to block the 4-OH group and thus convert it to an anti-directing one?

For this purpose, the isoxazolines $\frac{2}{2} - \frac{4}{2}$ were prepared from furoisoxazolines $\frac{1}{2}$ and $\frac{8}{2}$, in turn accessible in 60 - 70 % yield by nitrile oxide cycloadditions to furan and 2-methylfuran, respectively. ^{1,4,8} A solution of $\frac{1}{4}$ in methylene chloride/methanol (ca. 5 %)⁹ was treated with ozone at -78°C ¹⁰ until complete conversion was indicated by TLC control (ca. 10 min with 10 mmol runs). Dimethylsulfide was added to destroy hydroperoxides and ozonides. The resulting yellow oil (intractable NMR spectra) was transformed in 47 % yield to the crystal-line acetal $\frac{2}{2}$ ¹¹ (with neopentylglycol in refluxing benzene and TsOH catalyst), after NaOH treatment to cleave 0-acyl groups and prevent the detrimental elimination to isoxazoles. On the other hand, the crude ozonation product was reduced with NaBH₄ in methanol/water to produce $\frac{3}{2}$ (53 %) as a syrup, devoid of carbonyl absorptions (1R). The methylfuran adduct $\underline{1}\underline{b}$ gave slightly better yields and was used therefore in subsequent experiments. Crude $\underline{3}$ (from $\underline{1}\underline{b}$, ca. 60 %) was converted by azeotropic removal of methanol to the isopropylidene derivative $\underline{4}$ with dimethoxypropane/p-toluenesulfonic acid (catalyst) in benzene. Filtration through basic aluminia gave pure $\underline{4}$ (31 %, yellow oil ¹¹).

The isoxazolines 2 - 4 were reduced with LiAlH₄ and the crude products analyzed by ¹³C NMR (100.6 MHz) as usual. 5^{-7} From 3 a mixture of aminodeoxypentitols was isolated which, after peracetylation to furnish 6, consisted of two diastereoisomers in a 58 : 42 ratio (by NMR). However, configurational assignments (<u>xylo/arabino</u>) were not possible on the basis of NMR data. ¹¹ In contrast, 2 and 4 afforded on reduction a single diastereomer (d.r. > 97 : 3) in each case, in 95 and 73 % yield, respectively. The spectroscopic data of both 5 and 7 were again inconclusive in regard to their relative configurations. ¹¹

An analogous situation was met with $\underline{9}$ which was prepared in 34 % over-all yield from the furoisoxazoline acetal $\underline{8}^{1,4}$ by the sequence ozonolysis, NaBH₄ reduction, transacetalization of the diethyl acetal group with neopentyl glycol, and acetonide formation.¹¹⁻¹³ The reduction led to a single diastereomer (> 97 : 3) of the corresponding pentosamine derivative $\underline{10}$, as expected from the results for $\underline{4} \rightarrow \underline{7}$, in 88 % yield. The coupling constants¹¹ show $\underline{10}$ and $\underline{7}$ to belong to the same diastereomer series; these NMR data, however, did once more not allow to establish which one!

X-ray analysis was therefore performed on the structurally complementary aminodeoxy-pentoses $\frac{5}{2}$ and $\frac{10}{2}$ to elucidate the course of these highly stereoselective reductions. Single crystals of $\frac{5}{2}$ were prepared by repeated crystallization¹⁴ and suitable material from $\frac{10}{2}$ was obtained in form of the 0,N-diacetyl derivative.¹⁵ The 4-amino-4-deoxy-pentose $\frac{5}{2}$ proved to be the arabino isomer and the 2-amino-2-deoxypentose $\frac{10}{2}$ turned out the xy/o isomer!

The low selectivity in the reduction of $\frac{3}{2}$, an isoxazoline with two groups individually favouring <u>syn</u>-addition,⁷ is not expected and may be rationalized by the formation of counteracting <u>cyclic</u> aluminates. On the other hand, the strong anchoring effect of the 4-hydroxy group in $\frac{2}{2}$ to guide hydride addition in LAH reductions is <u>not</u> side-tracked by the <u>cis</u>-5dioxanyl substituent.¹⁶ In addition, the stereochemical outcome of the reductions of the bicyclic isoxazolines $\frac{4}{2}$ and $\frac{9}{2}$, to produce <u>xylo</u> derivatives $\frac{7}{2}$ and $\frac{10}{2}$ only, exceeds σ priori estimates based on results with related substrates.¹⁷

The results of this and previous³ studies show that stereo **control** in isoxazoline reductions can be achieved, based upon the presence of a free or blocked hydroxy group. The underlying principle may be generally useful in synthesis.¹⁸ Specifically, the syntheses of 5, 7, and 10 - derivatives of complementary structure and configuration - provide novel examples for the utilization of the isoxazoline route¹⁹ in amino sugar synthesis.²⁰

Acknowledgements: Support of this work by Deutsche Forschungsgemeinschaft, Fonds der Chemischen Industrie, Bayer AG (Dr. H. Meyer, Wuppertal), and Hoechst AG (Dr. W. Bartmann, Hoechst), is gratefully acknowledged. We are obliged to Ms. D. Michel for experimental help.

2998



References and Notes

- Syntheses via Isoxazolines, 17; (a) part 16: V. Jäger, I. Müller, <u>Tetrahedron</u>, in press;
 (b) part of the <u>Dissertation</u> of I. M., Würzburg 1984, (c) presented first at the <u>3rd</u> <u>Convegno Sulle Reazioni Pericicliche</u>, Naples, Oct. 1984.
- 2. V. Jäger, R. Schohe, Tetrahedron 40, 2199 (1984) and references therein.
- 3. W. Schwab, V. Jäger, Angew. Chem. 93, 578 (1981); Ibid., Int. Ed. Engl. 20, 603 (1981).
- 4. I. Müller, V. Jäger, Tetrahedron Letters 4777 (1982).
- Review: V. Jäger, H. Grund, V. Buß, W. Schwab, I. Müller, R. Schohe, R. Franz, R. Ehrler, Bull. Soc. Chim. Belg. 92, 1039 (1983).
- V. Jäger, V. Buß, W. Schwab, <u>Tetrahedron Letters</u> 3133 (1978); V. Jäger, V. Buß, <u>Liebigs</u> <u>Ann. Chem.</u> 101 (1980); V. Jäger, V. Buß, W. Schwab, <u>Ibid</u>. 122 (1980).
- 7. V. Jäger, W. Schwab, V. Buß, <u>Angew. Chem.</u> 93, 576 (1981); <u>Ibid, Int. Ed. Engl.</u> 20, 601 (1981).
- Furan additions to aryl nitrile oxides: ref. 1a and P. Grünanger, I. Grasso, <u>Gazz. Chim.</u> <u>Ital. 85</u>, 1271 (1955); A. Corsico Coda, P. Grünanger, G. Veronesi, <u>Tetrahedron Letters</u>, 2911 (1966); P. Caramella, G. Cellerino, A. Corsico Coda, A. Gamba Invernizzi, P. Grünanger, K. N. Houk, F. Marinone Albini, <u>J. Org. Chem. 41</u>, 3349 (1976). Intramolecular case: O. Tsuge, K. Ueno, S. Kanemasa, <u>Chem. Lett.</u> 285 (1984).

- 9. P. S. Bailey, Ozonation in Organic Chemistry, Vol. 1, Olefinic Compounds, Academic Press. New York 1978.
- 10. Isoxazolines at -78 ^OC are rather reluctant towards ozone: cf. ref. 5; H. Grund, <u>Dissertation</u>, Gießen 1978; A. P. Kozikowski, M. Adamczyk, <u>Tetrahedron Letters</u> 3123 (1982).
- 11. All compounds mentioned were characterized by elemental analyses, IR, ¹H and ¹³C NMR spectra (CDCl₃ solutions). Some characteristic data are: 2: m.p. 112 113 ^oC (from ether/p.e.); $\delta = 4.28$ (dd, J = 5.5 and 7.5 Hz; 5-H) and 5.20 ppm (dd, broad; $J_{45} = 7.5$, $J_{4,OH} = 3.8$ Hz, 4-H). 3: $\delta = 5.25$ ppm (bd, $J_{45} = 8$ Hz; 4-H). 4: $\delta = 4.29$ (ddd, J = 4.5, 4.5 and 7.0 Hz; 5-H), 5.11 ppm (d, J = 7.0 Hz; 4-H). 5: m.p. 120 122 ^oC; $\delta = 3.25$ (ddd, $J_{34} = 5.0$, $J_{45} = 4.0$ and 7.5 Hz; 4-H), 3.78 (dd, $J_{12} = 4.0$, $J_{23} = 1.0$ Hz; 2-H), 3.83 (dd, $J_{23} = 1.0$, $J_{34} = 5.0$ Hz; 3-H). 6: 56 % crude, 42 % with d.r. 53: 47, oil. One diastereomer was obtained pure by fractional crystallization, m.p. 133 134 ^oC. 7: light yellow oil; $\delta = 3.06$ (ddd, $J_{12} = 6.5$ and 6.5, $J_{23} = 3.0$ Hz; 2-H). 9: m.p. 133 134 ^oC; $\delta = 2.98$ (dd, $J_{12} = 6.5$, $J_{23} = 2.0$ Hz; 2-H). 2-N,4-O-diacetyl derivative (with Ac₂O, pyridine/DMAP, 2d at room temp.; 60 % after crystallization from hexane): m.p. 142 143 ^oC; $\delta = 4.19$ ppm (ddd, $J_{12} = 3.0$, $J_{23} = 4.0$, $J_{23} = 1$ Hz).
- 12. The transacetalization proved neccessary, as the diethyl acetal did not survive acid-catalyzed acetonide formation. The diethyl acetal group was chosen for the starting isoxazoline 8 in view of the considerably higher lability of cyclic acetals towards ozone.¹³
- 13. P. Deslongchamps, P. Atlani, D. Frehel, A. Malaval, C. Moreau, Can. J. Chem. 52, 3651 (1974).
- 14. X-ray of $\frac{5}{2}$: single crystals, m.p. 128 °C; 8 crystallizations from heptane. a = 6.286 (1), b = 10.620 (1), c = 13.865 (2) Å, a = 71.82 (1), B = 76.94 (1), γ = 79.30 (1) °; P1; Z = 2; D_x = 1.138, D_m = 1.13 g/cm³; R_1 = 0.12, R_2 = 0.045 (R3-Nicolet diffractometer; 20/0 scan; MoK_a; ϑ_{max} = 22.5 °; 1679 from 2287 independent reflexions; F > 1 σ (F); w = 1/ σ ² (F). Program system: G. M. Sheldrick, SHELXTL. Göttingen 1983; standard deviation of CC bond lengths: 0.007 Å.
- 15. X-ray of 10: single crystals, m.p. 144 °C; 5 crystallizations from heptane. a = 6.180 (1), b = 11.125 (1), c = 14.686 (2) Å, $\alpha = 76.56$ (1), $\beta = 85.50$ (1), $\gamma = 89.26$ (2) °, $P\overline{1}$; Z = 2, $D_x = 1.219$, $D_m = 1.22 \text{ g/cm}^3$; $R_1 = 0.057$, $R_2 = 0.033$ (2253 from 2563 independent reflexions); standard deviation of CC bond lengths: 0.004 Å.
- 16. This was confirmed independently: LAH reduction of a 5-dioxanyl isoxazoline led to a 54: 46 mixture of corresponding amino alcohols W. Kütt, <u>Diploma Thesis</u>, Würzburg 1984.
- 17. Cf. reductions of furoisoxazolines ^{1,4} and of cyclopentane-annulated isoxazolines.⁶
- For a recent case of OH-directed catalytic hydrogenation see (a) G. N. Stork, D. E.
 Kahne, J. Am. Chem. Soc. 105. 1072 (1983), (b) ref. 7 and references given therein.
- 19. V. Jäger, H. Grund, Angew. Chem. 88, 27 (1976); Ibid., Int. Ed. Engl. 15, 50 (1976).
- 20. For syntheses of 4-amino-4-deoxypentoses (xylo/lyxo) see H. Paulsen, K. Propp, J.Brüning, Chem. Ber. 102, 469 (1969); H. Paulsen, J. Brüning, K. Heyns, Chem. Ber. 103, 1621 (1970).

(Received in Germany 29 March 1985)